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subdiscipline of organometallic  
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coordination compounds, which  
are made up of metal ions  
surrounded by other non-metal  
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Hexagonal planar Geometry:

Unknown for first row transition metal ions, although the arrangement of six groups in a plane is found in some higher coordination number geometries..

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Trigonal prism Geometry: Most trigonal prismatic compounds have three bidentate ligands such as dithiolates or oxalates and few are known for first row transition metal ions.

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