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1s²2s²2p⁶3s²3p² P

1s²2s²2p⁶3s²3p³

26 24 S 1s²2s² 2P

3s²3p⁴ Cl

1s²2s²2p⁶3s²3p⁵

Ar

1s²2s²2p⁶3s²3p⁶

1 1 4 2 59. 4 62. 2

65. 2 68. 2 The

mixture contains at

least three parts.

One solid remained

on the filter paper.

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within the same
group of the period
table behave

similarly because
they have the same
number of what?

(octet rule, valence
electrons, electron
dot structures,
cations, group)

Valence electrons.

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How many valence
electrons does each
carbon atom have?

2. What gives
carbon the ability to
form chains that are
almost unlimited in
length?

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