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neutrons and six electrons (c) (i) State what is meant by the term relative atomic mass. (2)

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Question number Answer Mark 4 The only correct answer is D (3p subshell 6, third quantum shell 18) A is incorrect

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because 2 is the number of electrons in a 3p orbital and the 3d electrons have been omitted from the third quantum shell
B is incorrect because 2 is the number of electrons in a 3p orbital
C is incorrect because the 3d electrons have been omitted from the third quantum shell

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and neutrons, surrounded by electrons, orbiting in shells 1.10
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