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Reaction Net

periodiek

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Dot Structures

Het

wetenschappelijk

liefdeslied

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~~8: Section 1:~~

~~Molecular~~

~~Compounds~~

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7: Section 1:

Ions Chapter 9

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~~The Periodic~~

~~Table: Atomic~~

~~Radius,~~

~~Ionization~~

~~Energy, and Elec~~

~~tronegativity **AP**~~

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Equilibrium! *How
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by. Christine-
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Section 2.

Section 3. Terms
in this set (7)

mole. The SI
base unit used
to measure the

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substance whose
number of
particles is the
same as the
number of atoms
of carbon in
exactly 12 grams
of carbon ...

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mole. The amount
of a substance
that contains as
many particles
as there are
atoms in exactly
12 grams of

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Avogadro's
number.

6.022×10^{23} , the
number of atoms
or molecules in
1.000 mol. molar
mass. the mass
in grams of one
mole of a
substance.

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