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20 ° C. She then mixes 500 g of copper at 500 ° C with 1,200 g of water at 20 ° C. The specific heat capacity of lead is 0.1276 J/g ° C and the specific heat capacity of copper is 0.3845 J/g ° C.



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**L**ab Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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**Gizmo Warm-up** A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance ' s specific heat capacity.

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