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electron configurations | The Cavalcade o' Chemistry
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Worksheet Part 1. For each element, write the ground state
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Write the abbreviated ground state electron configurations for the following: 8) helium $1s^2$ (this one cannot be abbreviated) 9) nitrogen $[\text{He}] 2s^2 2p^3$ 10) chlorine $[\text{Ne}] 3s^2 3p^5$ 11) iron $[\text{Ar}] 4s^2 3d^6$ 12) zinc $[\text{Ar}] 4s^2 3d^{10}$ 13) barium $[\text{Xe}] 6s^2$ 14) polonium $[\text{Xe}] 6s^2 4f^{14} 5d^{10} 6p^4$

Electron Configurations Worksheet - iMater

Electron Configurations - Solutions Note: The electron configurations in this worksheet assume that lanthanum (La) is the first element in the 4f block and that actinium (Ac) is the first element in the 5f block. If your periodic table doesn't agree with this, your answers for elements near the f-orbitals may be slightly different. 1) sodium

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Let's learn all about the particles that make up the atom. The electron gets a lot of discussion, but I think you'll be happy with how much attention protons and neutrons get! (Updated 2/24/16) Protons, neutrons, and electrons: These guys live in the atom and have crazy parties. Or something like that. Anyway, learn about...

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