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Solution of a salt of the metal to be coated; and the anode (positive electrode) is usually either a block ...

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Chapter 2 Problem 2.19 1/1 Determine the solubility product K_{sp} for $PbSO_4$.

The desired equilibrium is $PbSO_4$

$Pb^{2+} + SO_4^{2-}$ We can write this as the sum of two electrochemical equations



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