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the change in the amount of reactants or  
products per unit time 1st Order:  $\text{Rate} = k[X]^1$   $\square$  Constant Half Lives 2nd Order:  $\square$   
 $\text{Rate} = k[Y]^2$   $\square$  Increasing Half Lives 0  
Order:  $\square$   $\text{Rate} = k[Z]^0$  Continuous Rate  
Method Initial Rate Method 1o  
Halogenoalkane + Hydroxide = SN2 3o  
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