

## Chemistry Chapter 14

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Section 14.1 Collision Theory: A Model for the Reaction Process. Goals. To describe a model, called collision theory, that helps us to visualize the process of many chemical reactions. To use collision theory to explain why not all collisions between possible reactants lead to products. To use collision theory to explain why possible reactants must collide with an energy equal to or above a certain amount to have the possibility of reacting and forming products.

### Chapter 14 - The Process of Chemical Reactions

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NCERT INTENT QUESTIONS. 14.1. Glucose or sucrose are soluble in water but cyclohexane and benzene (simple six membered ring compounds) are insoluble in water Explain. Ans: The solubility of a solute in a given solvent follows the rule ' Like dissolves like'. Glucose contains five and sucrose contains eight -OH groups.

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### CHAPTER 14: An Introduction to Organic Chemistry

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