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[PREFACE](#) The sixth edition of General Chemistry: The Essential Concepts, continues the tradition of presenting only the material that is essential to a one-year general chemistry course. As with previous editions, it includes all the core topics that are necessary for a solid foundation in general chemistry without sacrificing depth, clarity, or rigor.

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[Introductory Chemistry: Concepts and Critical Thinking, 6th Edition](#) © 2011 Pearson Education, Inc. Charles H. Corwin. EXAMPLE EXERCISE 3.2. Metric Unit Equations. Continued. Answers: (a) 1 Gm = 1 ×10. 9. m (b) 1 g = 1 ×10. 2. cg (c) 1 L = 1 ×10. 6. µL (d) 1 s = 1 ×10. 3. ms. Complete the unit equation for each of the following exact metric equivalents:

[EXAMPLE EXERCISE 3.1 Metric Basic Units and Prefixes](#)

Although the sixth edition incorporates many impressive features, such as macro to micro artwork, animations correlated to the text, and hand-sketched worked examples, General Chemistry is still 200 to 300 pages shorter and much less expensive than other two-semester textbooks. Dr.

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Description More than a textbook, Introductory Chemistry: Concepts and Critical Thinking, Sixth Edition is a comprehensive learning system that offers print and media resources as well as an extensive study area available in MasteringChemistry®.

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