

Chapter 9 Test Chemistry

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Chapter 9 - Ionic and Covalent Bonding . 1. In which pair do both compounds exhibit predominantly ionic bonding? A) RbCl and CaO B) PCl5 and HF C) KI and O3 D) Na2SO3 and BH3 E) NaF and H2O . ANS: A PTS: 1 DIF: easy REF: 9.1 OBJ: Define ionic bond. TOP: bonding | ionic bonding MSC: general chemistry . 2.

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Chapter 9 Practice Test - Naming and Writing Chemical Formulas Answer Section MATCHING 1. ANS: A PTS: 1 DIF: L1 REF: p. 253 OBJ: 9.1.1 Identify the charges of monatomic ions by using the periodic table, and name the ions. STA: Ch.5.a 2.ANS: H PTS: 1 DIF: L1 REF: p. 254

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This chapter is very important chapters for your chemistry notes for class 9. This chapter is about the bonding of the atoms and types of bonding. The most interesting question is why noble gases cannot form a chemical bond? Because that is stable. Every person wants to be rich, the same goes for the atom too. [lock]

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PRENTICE HALL CHAPTER 9 CHEMISTRY TEST. 5. The metals in Group 1A, 2A and 3A: Circle Answer gain electrons when they form ions