

Chapter 17 Ap Chemistry Review

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$\Delta H_o = 67.7 \text{ kJ}$ $\text{N}_2(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{N}_2\text{O}_4(\text{g})$ $\Delta H_o = 9.7 \text{ kJ}$ 13. Chapter 17 Review How much heat is required to raise the temperature of $6.5 \times 10^2 \text{ g}$ of aluminum by $30 \text{ }^\circ\text{C}$?

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