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atom is made up
of 3 parts,
protons,

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neutrons and electrons. All known elements have been arranged on the table of periodic elements which shows shared patterns and relationships based on the columns and rows they are aligned

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Structure: Atoms
consist of
protons and
neutrons in the
nucleus,
surrounded by
electrons that

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reside in
orbitals.
Orbitals are
classified
according to the
four quantum
numbers that
represent any
one particular
orbital energy,
shape,
orientation, and
the spin of the
occupying

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electron.
Periodic Table:
This is a
tabular
arrangement of
the chemical
elements,
ordered...

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Elements are
arranged on the
Periodic table

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in order of increasing atomic number where each element has one proton more than the element preceding it. Hydrogen has 1 proton, helium has 2 protons, lithium has 3 etc. The table is arranged in

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vertical columns
called Groups
numbered I –
VIII and in rows
called Periods.

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on the periodic
table is a
function of
their atomic
structure. The
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is a tool used
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or groups; e)
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In the periodic
table, chemical
elements are
arranged by
order of atomic
number in such a
way that the

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properties
Structure And
(chemical
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the elements are
made clear. The
standard form of
the table
includes periods
(usually
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table) and
groups (usually

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vertical).

Elements in
groups have some
similar

properties.

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indicate that -
f A is more
reactive than B
g A and B are
isotopes of the
same element h A
and B are
different
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based on the columns and rows they are aligned in. Each element on the periodic table has a unique atomic number which represents the number of protons the element has.

~~03.01 Atomic~~

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