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2.4 moles of ideal monoatomic gas starts at
state a with pressure 1 atm (1.01×10^5
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3200 N is equal to $3200/800 = 4 \text{ ms}^{-2}$ 2 2 a
force = mass*acceleration = $0.80 \times 20 = 16 \text{ N}$ 2 b
Mass = force/acceleration = $200/5 = 40 \text{ kg}$. 2
c Acceleration = force/mass = $840/70 = 12 \text{ m/s}^2$
2 . 2 d Force = mass*acceleration = $0.40 \times 6 =$

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2.4N. 2 e Mass = force/acceleration =
 $5000/0.20 = 25000\text{kg}.$

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$a = (V^2/R - cv/R \cos(v)) + (dv/dt + cv/R \sin(v) + V^2/R r)$ where $R=ct$ and $dR=cdt$.
 $cv/R\cos(v)$ is the Centrifugal Acceleration a part of the real accelerations in the first parenthesis.

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