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This is a collection of worked general chemistry and introductory chemistry problems, listed in alphabetical order. Included are printable pdf chemistry worksheets so you can practice problems and then check your answers. You may also browse chemistry problems according to the type of problem.

*Practice Chemistry with Worked Chemistry Problems*

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Ksp Problems – Chemistry Name: \_\_\_\_\_ 1) The value of Ksp of AgCl is  $1.8 \times 10^{-10}$ . What would be the molar concentration of Ag<sup>+</sup> and Cl<sup>-</sup> in pure water placed in contact with solid AgCl(s)?  $K_{sp} = [Ag^+][Cl^-]$   $1.8 \times 10^{-10} = [x][x]$   $x = 0.000013416$  M or  $1.3416 \times 10^{-5}$  M 2) The molar solubility of PbCl ...

*Ksp Problems – Chemistry*

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3—3 Practice Problems 1. How many protons and electrons are 8. What is the name of element that has present in a vanadium atom? atoms that contain 25 protons? 2. How many protons and electrons are 9. What is the name of the element that has present in a nitrogen atom? atoms that contain 82 protons? 3. How many protons and electrons are 10.

*Name Date Class 3—3 Practice Problems - Weebly*

0.710 kg of sodium carbonate (washing soda), Na<sub>2</sub>CO<sub>3</sub>, in 10.0 kg of water—a saturated solution at 0 °C; 125 g of NH<sub>4</sub>NO<sub>3</sub> in 275 g of water—a mixture used to make an instant ice pack; 25 g of Cl<sub>2</sub> in 125 g of dichloromethane, CH<sub>2</sub>Cl<sub>2</sub>; 0.372 g of histamine, C<sub>5</sub>H<sub>9</sub>N, in

125 g of chloroform,  $\text{CHCl}_3$

*8.3: Concentrations of Solutions (Problems) - Chemistry ...*

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